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# CO<sub>2</sub>-PROCESSES PHOTORESISTS, POLYMERS, AND PHOTOACTIVE COMPOUNDS FOR MICROLITHOGRAPHY

5 <u>Cross-Reference to Related Applications</u>

The present application claims priority to Provisional Application Nos. 60/239,749, filed October 12, 2000 and 60/267,993 filed February 9, 2001, the disclosures of which are incorporated herein by reference in their entirety.

10 <u>Field of the Invention</u>

The invention generally relates to processes relating to forming microelectronic devices using carbon dioxide.

#### Background of the Invention

There are a number of lithography processes which are well known in the art used in conjunction with manufacturing microelectronic devices. A typical lithography process involves aligning and transferring a pattern from a template using optics onto a partially processed substrate (e.g., wafer) that has been coated with a photoresist. The resist is then developed and the resist image is thereafter transferred into the underlying material typically by a chemical and/or thermal step such as, for example, dry/wet etch, sinter, implant, etc. This process is often repeated many times (e.g., between 15 to 20) during the course of building a complex integrated circuit.

The above processing steps can result in distortions being present in the wafer. Thus, wafer handling during processing is extremely important.

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The wafer handling subsystem is believed to be largely responsible for the throughput of the processing tool. Accordingly, the wafer handling should be designed to minimize sources of possible contamination. A difficulty from a processing standpoint relates to the incompatibility of an organic or aqueous solvent based coating solution and the enclosed equipment (e.g., a vacuum environment) used for processing substrates that have developed photo resist images (positive or negative) formed therein.

U.S. Patent No. 5,665,527 to Allen et al. proposes a process for generating a negative tone resist image comprising the steps of coating a substrate with a film of a polymeric composition comprising a polymer, a photosensitive acid generator, and acid labile groups, imagewise exposing the film to radiation to generate free acid, and developing the image with critical fluid, e.g., carbon dioxide.

Notwithstanding the above, there is a need in the art for processes for forming semiconductor devices that involve depositing, developing and stripping resists that may be carried out in an integrated closed system which uses a solvent which is relatively benign from an environmental perspective (e.g., carbon dioxide). Such a process would be highly advantageous in that it would reduce or obviate the repeated entry, removal, and re-entry steps that are inherent in non-integrated processes.

#### Summary of the Invention

In one aspect, the invention provides a process of forming a resist image in a microelectronic substrate. The process comprises the steps of contacting the substrate with a first composition comprising carbon dioxide and a component selected from the group consisting of at least one polymeric material, at least one monomer, at least one polymeric precursor, and mixtures thereof to deposit the component on the substrate and form a coating thereon, then imagewise exposing the coating to radiation such that exposed and unexposed coating portions are formed, and then subjecting the coating to a second composition comprising carbon dioxide such that either one of the exposed or the unexposed coating portions are removed from the

substrate and the other coating portion remains on the coating to form an image thereon (e.g., developing the image using carbon dioxide). For the purposes of the invention, the first and second compositions may be the same or different. In a preferred embodiment, each step may be performed in a closed system such that said process is an integrated process. For the purposes of the invention, the term "closed system" refers to a system that is not exposed to ambient conditions.

This and other aspects are described by the invention in greater detail herein.

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#### <u>Detailed Description of the Preferred Embodiments</u>

The invention will now be described in greater detail in reference to its preferred embodiments described in the text below and in the drawings. It should be understood that these embodiments are for illustrative purposes only, and do not limit the scope of the invention as defined by the claims.

Stated alternatively, the invention encompasses a method of forming a patterned resist layer on a substrate comprising the steps of forming on the substrate a resist layer comprising a material selected from the group consisting of at least one polymeric material, at least one monomer, at least one polymeric precursor, and mixtures thereof, by contacting the substrate with a first composition comprising the material and carbon dioxide, developing portions of the resist layer, and selectively removing developed or undeveloped portions of the resist layer by exposing the resist layer to a second composition comprising carbon dioxide.

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Various substrates can be used for the purposes of the invention such as, for example, substrates which may comprise any number of materials including, but not limited to, silicon, ceramics, polymer, gallium arsenide, silicon carbide, or the like, and combinations thereof. In various optional embodiments, at least one intermediate layer (e.g., a silicon dioxide layer) may be present between the substrate and the coating.

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The step of contacting the substrate which results in the formation of a coating thereon may be carried out using various techniques. Examples of

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embodiments include, without limitation, a spin coating method, a dip coating method, and a spray coating method. A spin coating method is preferred, and is described in U.S. Patent No. 6,001,418 to DeSimone et al., the disclosure of which is incorporated herein by reference in its entirety. A meniscus coating technique can also be employed such as one described, for example, in U.S. Patent No. 6,083,565 to Carbonell et al., the disclosure of which is incorporated herein by reference in its entirety. A coating method using self-assembling monolayers (SAMs) can also be employed, and may be used in conjunction with CO<sub>2</sub>. Examples of molecules that are capable of producing SAMs include, without limitation, alkane thiols, fluorinated alkane thiols, and functionalized alkane thiols such as those represented by the formula:

 $X-(CH_2)_n-S-H$  wherein X may be -CN, -COOH,  $-OCH_3$ ,  $-(OCH_2CH_2)_n-OH$ ,  $-(OCH_2CH_2)_n-OCH_3$ , -H, phenyl, or substituted phenyl, and n ranges from 1 to 1,000, and more preferably from 2 to 20.

A wide range of components can be used in the present invention to form a coating on the substrate. For the purposes of the invention, the term "materials" can include, without limitation, precursors, monomers, polymeric materials, and the like. Polymers, polymeric precursors, and monomers that are soluble or insoluble in CO2 or CO2 mixed with other gases (insert or otherwise) and co-solvents including surfactants may be necessary to maintain low surface tension and solubility and viscosity as CO<sub>2</sub> evaporates, at a number of density and/or pressure conditions are encompassed by the invention. Examples of suitable embodiments include polymeric materials which comprise, without limitation, fluoropolymers, silicon-containing polymers, or combinations thereof, such polymers having a CO<sub>2</sub>-soluble or "CO<sub>2</sub>-philic" component. "CO<sub>2</sub>-philic" refers to the polymer having an affinity for carbon dioxide. Examples of suitable monomers for making the fluoropolymers are numerous and include, but are not limited to, those set forth in U.S. Patent No. 5,739,223 to DeSimone et al., the disclosure of which is incorporated herein by reference in its entirety. Carbon dioxide soluble monomers may be employed if so desired. Exemplary monomers are

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fluoroacrylate monomers, fluorostyrene monomers, fluoroalkylene oxide monomers, fluorolefin monomers, fluorinated alkyl vinyl ether monomers, cyclic fluorinated monomers, and mixtures thereof. Specific preferred monomers include, without limitation, 2-(N-ethylperfluorooctane- sulfonamido) ethyl acrylate, 2-(N-ethylperfluorooctane- sulfonamido) ethyl methacrylate, 2-(N-methylperfluorooctane- sulfonamido) ethyl acrylate, 2-(Nmethylperfluorooctane- sulfonamido) ethyl methacrylate, 1,1'dihydroperfluorooctyl acrylate, 1,1'-dihydroperfluorooctyl methacrylate, 1,1',2,2'-tetrahydroperfluoroalkylacrylate, 1,1',2,2'-tetrahydroperfluoroalkyl methacrylate, α-fluorostyrene, 2,4,6-trifluoromethylstyrene, hexafluoropropylene oxide, perfluorocyclohexane oxide, tetrafluoroethylene, vinylidine fluoride, chlorotrifluoroethylene, perfluoro(propyl vinyl ether), perfluoro(methyl vinyl ether), 2,2-bis-trifluoromethyl-4,5-difluoro-1,3-dioxole, and mixtures thereof, including those set forth in U.S. Patent No. 6,083,565 to Carbonell et al., the disclosure of which is incorporated herein in its entirety by reference.

Any number of silicon-containing polymers can be used in the present invention, the selection of which is known by those skilled in the art. Examples of silicon-containing polymers include those containing at least one segment such as, without limitation, an alkyl siloxane, a fluoroalkyl siloxane, a chloroalkyl siloxane, and mixtures thereof.

Various polymeric materials that may be used in forming the coating (e.g., photoresist resin) include, without limitation, a polymer resin formed from a t-butyl horbornyl group, maleic anhydride, a fluorinated norbornyl group, and acrylic acid. Such a resin is particularly effective at a radiation of 193 nm wavelength. Another example of a suitable resin is an O<sub>2</sub>-RIE (Reactive Ion Etch) resistant random terpolymer of 1,1'-dihydroperfluorooctyl methacrylate ("FOMA"), t-butyl methacrylate ("TBM"), and a poly(dimethylsiloxane) macromonomer such as, for example, CH2=C(CH<sub>3</sub>)COOCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-(Si(CH<sub>3</sub>)<sub>2</sub>O)<sub>n</sub>-SI(CH<sub>3</sub>)<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub> wherein n ranges from 1 to 10,000.

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Monomers that may be used include, without limitation, carbon dioxide soluble monomers such as those fluoro- and silicon monomers described herein, as well as (meth)acrylate monomers (e.g., methyl (meth)acrylate) and aryl vinyl monomers (e.g., styrene). These monomers, as well as polymeric precursors, may be deposited as a thin film on the substrate, and polymerized thereon using any suitable technique such as, for example, thermal or photochemical means, including those described in U.S. Patent No. 6,083,565 to Carbonell et al.,.

As described above, the coating is imagewise exposed to radiation as described in U.S. Patent No. 5,665,527 to Allen et al., the disclosure of which is incorporated herein by reference in its entirety. Various radiation techniques including electromagnetic radiation such as deep ultraviolet or X-ray, more preferably deep ultraviolet radiation suitably at a wavelength of about 150-250 nm (248/254 nm), preferably 157 and 193 nm, and e-beam. Suitable radiation sources include, without limitation, mercury, mercury/xenon, xenon lamps, and X-ray. Visible radiation can also be employed. A typical radiation source is ArF excimer or KrF excimer.

The invention may optionally include other embodiments. In one aspect for example, the invention may further comprise the steps of depositing a metal-containing material or an ionic material on the surface of the substrate from which the exposed or the unexposed coating portions was removed, and removing the remaining (exposed or unexposed) coating portion from the substrate. The step of depositing a metal-containing material or an ionic material can be carried out (e.g., processed) by any number of methods including, without limitation, ion implant, metal deposition, and the like. Metal depositions techniques include, without limitation, evaporation sputtering, chemical vapor deposition, or laser-induced deposition. Suitable metal-containing materials include conductive metals such as, but not limited to, aluminum, copper, gold, titanium, tantalum, tungsten, molybdenum, silver, combinations thereof, and alloys thereof. Suitable ionic materials include, but are not limited to, inorganic ions such as boron, phosphorous, or arsenic which can be implanted. Accordingly, n- or p-doped transistors can be

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formed. In one preferred embodiment, the method further comprises removing the exposed coating portion from the substrate comprises contacting the exposed coating portion with a carbon dioxide containing fluid such that the exposed coating portion is removed from the substrate. In another preferred embodiment, the method comprises removing the unexposed coating portion from the substrate comprises contacting the unexposed coating portion with a carbon dioxide containing fluid such that the unexposed coating portion is removed from the substrate.

In one embodiment of the invention, the exposed coating portion has a lower solubility in carbon dioxide relative to the unexposed coating portions. Accordingly, the step of subjecting the coating to a composition comprising carbon dioxide comprises removing the unexposed coating portion from the substrate such that the exposed coating portion remains. Thereafter, the exposed resist can be removed using carbon dioxide as set forth herein.

Other embodiments that may be encompassed by the invention include, for example, depositing one or more insulating layers (e.g., interlayer dielectrics (ILDs)) and/or one or more anti-reflective coatings on the substrate. These steps may be carried out in a manner appropriate with the other steps of the process of the invention. The ILDs may be used to keep signals in the microelectronic device from straying between the conducting lines on the chip. Examples of ILDs include, without limitation, materials with low dielectric constants such as SILK resins made commercially available by The Dow Chemical Company of Midland, Michigan (described in detail herein) and Teflon AF<sup>TM</sup> made commercially available by E.I. DuPont de Nemours of Wilmington, Delaware. Examples of materials for anti-reflective coatings include, without limitation, a fluorinated material based on a fluoroacrylate polymer of polymer precursor.

For the purposes of illustration, an embodiment for illustrating the phase diagram for a negative resist (i.e., base resin) before and after exposure to radiation is set forth in **FIG. 1**. Above each curve, the resin is soluble, below it is insoluble. In accordance with the invention, the resists (e.g., coatings) of the invention can be deposited using carbon dioxide and

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also removed using carbon dioxide such as, for example, by a "stripping" technique. For the purposes of the invention, the term "stripping" refers to the removal of remaining resist from a substrate (e.g., wafer) whether it has been exposed to radiation or not. An example of CO<sub>2</sub> conditions for providing an integrated negative resist system is as follows:

Deposition Density >  $\rho_1$  (e.g., liquid CO<sub>2</sub>)  $\rho_1$  < Development Density <  $\rho_2$  (e.g., liquid or supercritical CO<sub>2</sub>) Removal (e.g., strip) Density >  $\rho_2$  (e.g., liquid or supercritical CO<sub>2</sub>)

As alluded to above, carbon dioxide is employed in a liquid or supercritical phase. If liquid CO<sub>2</sub> is used, the temperature-employed during the process is preferably below 31°C. As used herein, "supercritical" means that a fluid medium is at a temperature that is sufficiently high that it cannot be liquefied by pressure. The thermodynamic properties of CO<sub>2</sub> are reported in Hyatt, *J. Org. Chem.* **49**: 5097-5101 (1984); therein, it is stated that the critical temperature of CO<sub>2</sub> is about 31°C.

Although not intending to be bound by theory, it is believed that in one embodiment the decrease in solubility in CO<sub>2</sub> of the resist polymer upon exposure to radiation can be achieved by the removal of a blocking group on the polymer which is soluble in carbon dioxide by using a photo acid generator (PAG) that exposes a polar group which renders the polymer insoluble in CO<sub>2</sub>. Such an embodiment is illustrated in **FIG. 2**. Examples of PAGs that may be used are numerous and conventionally known in the art including, without limitation, triaryl sulfonium salt, diarylidonium salt, and mixtures thereof. Also, PAGs of the formulas listed hereinbelow can also be used:

$$\left( \begin{array}{c} \\ \\ \end{array} \right) X^{+}SO_{3}^{-}(CF_{2})_{3}CF_{3}$$

wherein n is 2 or 3, and X is preferably I or S.

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Thus, by manipulating the composition of the polymeric coating (e.g., selecting various polar groups), one can design the polymer such that it is less soluble upon exposure to radiation than the base resin, while allowing the polymer to be dissolved in CO<sub>2</sub> at higher densities and pressures. Thus, one may be able to deposit the polymer using CO<sub>2</sub>, develop the negative tone image using CO<sub>2</sub>, and remove (e.g., strip) it using CO<sub>2</sub>.

In one embodiment, and in accordance with the invention, one is capable of designing the resist such that it can be deposited by utilizing conventional techniques, i.e., known solvents, then developed by employing a first fluid or composition comprising carbon dioxide at a given density, and then stripping the resist using a second fluid or composition comprising carbon dioxide, either in a liquid or supercritical state. The first and second fluids or compositions may be the same or different. Thus, it is believed that the invention is distinguished from U.S. Patent No. 5,665,527 to Allen et al.

In another embodiment, the exposed coating portion has a higher solubility in carbon dioxide relative to the unexposed coating portion, and wherein the step of subjecting the coating to a second composition comprising carbon dioxide comprises removing the exposed coating portion from the substrate such that the unexposed coating portion remains. Such an embodiment describes the formation of a positive tone resist, i.e., upon exposure to radiation, the resist becomes more soluble in carbon dioxide. An embodiment illustrating a positive tone resist is set forth in **FIG. 3**. In one example, the CO<sub>2</sub>-soluble base resist may have a certain percentage of labile polar groups which, upon exposure to radiation, are removed which enhances the solubility of the exposed coating portion relative to the non-exposed coating portion.

Various polymeric materials may be employed as a coating on the substrate including, without limitation, all of those recited herein. For the purposes of the invention, the term "polymer" encompasses homopolymer, copolymer, terpolymer, and the like. In one embodiment, the component employed in the invention encompasses a polymeric material which is selected from the group consisting of a copolymer of a fluoroacrylate and a

component selected from the group consisting of C(CH<sub>2</sub>OH)<sub>3</sub>, a sugar unit, and SiR<sub>3</sub> wherein R is a polar group selected from the group consisting of:

$$-CH_2CH_2CH_2OH$$
,  $-CH_2CH_2CH_2-C\equiv N$ , and

mixtures thereof.

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One specific embodiment of a polymer that may be used as a positive tone resist is a copolymer of fluoroacrylate having the following comonomer structures:

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$$-CH_2$$
 $-CH_2$ 
 $-CH_$ 

wherein R polar group is preferably selected from the group consisting of:

mixtures thereof.

In one embodiment, upon exposure to radiation, the polar group is removed from the unit, thus resulting in the formation of a carboxylic acid unit that has a higher solubility in CO<sub>2</sub> as illustrated by the following:

$$-(CH_2-C-)$$
 $C=0$ 
 $OH$ 

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Not intending to be bound by theory, another embodiment that can allow for an increase of the polymer solubility upon exposure to radiation is to employ a polymer that is capable of undergoing chain scission upon such exposure, e.g., an alternating copolymer between a fluorinated alkyl allyl ether with sulfur dioxide. Upon exposure to radiation (e.g., e-beam), scission of the main polymer chain results and thus a lowering of the polymer molecular weight. Therefore, the resulting polymer has a higher solubility in CO<sub>2</sub> relative to the parent base resist of higher molecular weight.

An example an embodiment of a generic phase diagram for a positive resist is given in **FIG. 4**. Above each curve, the polymer (e.g., resin) is soluble

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in CO<sub>2</sub>, below it is insoluble. In terms of fluid density, the CO<sub>2</sub> conditions for achieving an integrated positive resist system is as follows:

Depositing Density >  $\rho_1$   $\rho_2$  < Development Density <  $\rho_1$ Removal (e.g., Strip) Density >  $\rho_1$ 

In embodiments encompassing either the negative-tone or positive-tone resists, it is possible to incorporate a refractory element such as, for example, silicon, to enable dry pattern transfer steps like  $O_2$ -RIE particularly in the form of microphase separated morphologies that are capable of elevating the glass transition temperature ( $T_g$ ) of the continuous phase (see e.g., DeSimone et. al. *Macromolecules* **1991**, 24 5330-5339).

Any of the carbon dioxide-containing compositions may also include additional components, the selection of which is known to one skilled in the art. Exemplary components include, without limitation, aqueous and organic co-solvents, polymer modifiers, water, rheology modifiers, plasticizing agents, flame retardants, antibacterial agents, flame retardants, and viscosity reduction modifiers.

In another aspect, the invention provides a process of making integrated circuits. The process comprises the steps of:

- (a) inserting at least one wafer into an integrated microelectronics process device (IMPD); then
- (b) contacting the at least one wafer with a first composition comprising carbon dioxide and a polymeric material to deposit the polymeric material on the substrate and form a coating thereon; then
- (c) imagewise exposing the coating to radiation such that exposed and unexposed coating portions are formed; then
- (d) subjecting the coating to a second composition comprising carbon dioxide such that either one of the exposed or the unexposed coating portions are removed from the at least one and the other coating portion remains on the coating to form an image thereon (i.e, developing the image with carbon dioxide); then

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- (e) depositing a metal-containing material or an ionic material on the surface of the wafer from which the exposed or the unexposed coating portions were removed; then
- (f) removing the exposed or unexposed coating portion from the5 wafer; and then
  - (g) removing the at least one wafer from the IMPD.

Advantageously, steps (a) through (f) are all performed repeatedly in the IMPD without the wafer being removed from the IMPD. In a preferred embodiment (b) through (f) are repeated at least once without removing the at least one wafer from the IMPD. One example of an IMPD that can be employed is described in U.S. Patent No. 6,045,877 to Gleason et al., the disclosure of which is incorporated herein by reference in its entirety.

In another aspect, the invention encompasses a process of making integrated circuits. The process comprises the steps of:

- (a) inserting at least one wafer into an integrated microelectronics process device (IMPD); then
- (b) contacting the at least one wafer with a first composition and a component selected from the group consisting of at least one polymeric material, at least one polymeric precursor, and at least one monomer, and mixtures thereof to deposit the component on the wafer and form a coating thereon; then
- (c) imagewise exposing the coating to radiation such that exposed and unexposed coating portions are formed; then
- (d) subjecting the coating to a second composition comprising carbon dioxide such that either one of the exposed or the unexposed coating portions are removed from the at least one and the other coating portion remains on the coating to form an image thereon (i.e., developing the image with carbon dioxide); then
- (e) depositing a metal-containing material or an ionic material on the surface of the wafer from which the exposed or the unexposed coating portions are removed; then

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- (f) removing the exposed or unexposed coating portion from the wafer; and then
- (g) removing the at least one wafer from the IMPD; wherein said steps (a) through (f) are performed in the IMPD without the at least one wafer being removed from the IMPD.

In another aspect, the invention encompasses a process of making integrated circuits. The process comprises the steps of:

- (a) contacting at least one wafer with a first composition and a component selected from the group consisting of at least one polymeric material, at least one polymeric precursor, and at least one monomer, and mixtures thereof to deposit the component on the wafer and form a coating thereon; then
- (b) imagewise exposing the coating to radiation such that exposed and unexposed coating portions are formed; then
- (c) subjecting the coating to a second composition comprising carbon dioxide such that either one of the exposed or the unexposed coating portions are removed from the at least one and the other coating portion remains on the coating to form an image thereon (i.e., developing the image with carbon dioxide); then
- (d) depositing a metal-containing material or an ionic material on the surface of the wafer from which the exposed or the unexposed coating portions are removed; and then
- (e) removing the exposed or unexposed coating portion from the wafer.

wherein said steps (a) through (e) are performed in the IMPD without the at least one wafer being removed from an IMPD. In a preferred embodiment (a) through (e) are repeated at least once without removing the at least one wafer from an IMPD. In another preferred embodiment, a selective etching step such as those described herein may be employed subsequent to step (c) but prior to step (d) to selectively remove various materials that may be present on the substrate including, without limitation, those described herein.

The first and second compositions utilized in the above embodiments may be the same or different.

The above steps may be carried out employing any of the embodiments described hereinabove, as well as others. For example, the component on the wafer, preferably the polymeric material, may be deposited using a chemical vapor deposition (CVD) method. CVD methods are known in the art, and one example of a CVD method is described in U.S. Patent No. 6,045,877 to Gleason et al. A particularly preferred polymeric material is a fluoropolymer.

The fluid composition may comprise various materials such as, in one embodiment, carbon dioxide or a carbon dioxide/co-solvent mixture.

The present invention may be carried out by employing various processing equipment. In one embodiment, for example, a spin coater can be used to deposit polymeric material on the substrate. One example of a spin coater is described in U.S. Patent No. 6,001,418 to DeSimone et al. that has ports, sub-chambers, to accommodate imaging, CO<sub>2</sub> removal, vacuum development, and/or stripping techniques. This embodiment provides one example of an IMPD. Modifications from the above may be made without departing from the scope of the invention.

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The present invention is highly advantageous in that the deposition, development, and removal (e.g., stripping) of the resist used in processing microelectric substrates can be carried out in a closed system, in contrast to conventional solvent-based processes. Moreover, due to the environmentally benign nature of carbon dioxide, and CO<sub>2</sub> mixtures mentioned herein, and its relative ease of handling, one has the ability to design microelectronics processing equipment (e.g., metal deposition facilities, etch processes, and O<sub>2</sub>-Reactive Ion Etch (RIE) processes that are integrated with carbon dioxide based deposition, development, and stripping equipment. Moreover, post-processed carbon dioxide is typically easier to recover than organic or aqueous wastes. The low solvent viscosity of carbon dioxide and CO<sub>2</sub> solutions also makes it highly desirable for penetrating increasingly smaller gaps between features in microelectronic substrates. Such an integrated

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design of the microelectronics processing equipment can avoid the repeated entry, removal, and re-entry steps that are believed to be inherent in non-integrated processes. In one embodiment, for example, one could process a wafer repeatedly within one device without removing it in order to spin coat or develop a photoresist coating. Such an integrated system can therefore eliminate or significantly reduce the need for extensive and costly clean room facilities that are often required to reduce wafer contamination upon removal from closed systems.

In other embodiments, the processes of the invention may include other steps. For example, in embodiments where an intermediate layer is present between the coating portion and the substrate, the process may further comprise the step of selectively etching the intermediate layer using the developed coating portion as an etching mask. The intermediate layer may comprise many of the components set forth herein, the selection of which is known to one skilled in the art. In a preferred embodiment, the intermediate layer is present in the form of a planarization layer such as one comprising a novalac resin, or an interlayer dielectric resin such as one selected from the SiLK™ family of dielectric resins made commercially available from The Dow Chemical Company located in Midland, Michigan. Exemplary resins in the SiLK™ family include SiLK™ I Resin SiLK™ H Resin, and CYCLOTENE™\*

The etching may be carried out by employing various techniques known to one skilled in the art. For example, the step of selectively etching the intermediate layer comprises contacting the intermediate layer with a gas selected from the group consisting of oxygen, chlorine, fluorine, and mixtures thereof. It should be appreciated that the term "etching" is broadly defined and may include methods which completely or partially expose the underlying substrate. The etching step may be used in conjunction with any of the other steps described herein as deemed appropriate by one skilled in the art. For example, subsequent to etching, the exposed substrate may be contacted (e.g., showered) with metal-containing materials doped with an ionic material

including inorganic ions such as, without limitation, boron, arsenic, or phosphorous.

The present invention will now be described by the examples which follow. It should be understood that these examples merely illustrate the invention, and do not limit the scope of the invention as defined by the claims.

## Example 1

# Solubility Study of Polymer Resin

A composition comprising carbon dioxide, poly(FOMA-r-TBM), and a photoacid generator is exposed to radiation. The following illustrates the reaction which occurs resulting in the formation of poly(FOMA-r-MAA):

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The polymer contains 33 mol percent of TBA. Poly (FOMA-r-TBM) is determined to be soluble at 1500 psi as measured by a room temperature cloud point measurement.<sup>1</sup> Poly (FOMA-r-MAA) is determined to be insoluble at 5000 psi as measured by the cloud point measurement. Using 33 mol percent of TBA provides solubility contrast for spin coating, and development with carbon dioxide.

## Example 2

# **Solubility Study of Polymer Resin**

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The procedure according to Example 1 is followed except that 20 mol percent of TBM is used. Poly (FOMA-r-TBM) is determined to be soluble at 1350 psi as measured by a room temperature cloud point measurement. Poly (FOMA-r-MAA) is determined to be insoluble at 5000 psi as measured by the

<sup>&</sup>lt;sup>1</sup> 4 weight percent CO<sub>2</sub>

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cloud point measurement. Using 20 mol percent of TBA provides solubility contrast for spin coating, and development with carbon dioxide.

#### Example 2

## Solubility Study of Polymer Resin

The procedure according to Example 1 is followed except that 12 mol percent of TBM is used. Poly (FOMA-r-TBM) is determined to be soluble at 1260 psi as measured by a room temperature cloud point measurement. Poly (FOMA-r-MAA) is determined to be soluble at 3000 psi as measured by the cloud point measurement. Using 12 mol percent of TBA provides solubility contrast for spin coating, development, and stripping with carbon dioxide.

This invention encompasses methods of forming patterned resist layers on a variety of substrates (including but not limited to silicon, ceramics, polymer, gallium arsenide, silicon carbide, glass, polycarbonate, quartz, poly(tetrafluoroethylene), gold). These resist layers include at least one monomer, a polymer, a photoactive compound and carbon dioxide as a casting solvent and/or processing solvent.

Examples of methods to cast are spin coating (e.g., as described in U.S. Patent No. 6,001,418, the disclosure of which is incorporated herein by reference in its entirety), free-meniscus coating (e.g., as described in U.S. Patent No. 6,083,565, the disclosure of which is incorporated herein by reference in its entirety), dip coating, spray coating, and self-assembling monolayers including those described hereinabove.

The materials for coatings include, without limitation, precursors, monomers, polymers, and oligomers such that these materials are soluble in CO<sub>2</sub>, as well as CO<sub>2</sub> in combination with inert gases, co-solvents, and CO<sub>2</sub>-soluble surfactants, as well as other gases that can control viscosity, surface tension, and density of the liquid phase. "CO<sub>2</sub>-philic" include those set forth herein as well as, without limitation, perfluoroalkyl substituted norbornene monomers, dinorbornene, and alicyclic monomers that increase etch resistance and Tg of the polymers. Siloxane polymers may be employed, as well as coating materials.

Other examples of resist materials are fluorinated resists include those described in WO 00/67072, the disclosure of which is incorporated herein by reference in its entirety. These resists may include, as an example appropriate components, e.g., photoactive components including, without limitation, those described herein as deemed necessary.

In certain embodiments, resists comprise a fluorine-containing polymer comprising a repeat unit derived from at least one ethylenically unsaturated compound containing a fluoroalcohol functional group having the structure:

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## -C (R<sub>f</sub>) (R<sub>f</sub>) OH

wherein  $R_f$  and  $R_f$  are the same or different fluoroalkyl groups of from 1 to about 10 carbon atoms or taken together are  $(CF_2)_n$  wherein n is 2 to 10. The fluorine-containing polymer has an absorption coefficient of less than 4.0 pm~l at a wavelength of 157 nm.

In another embodiment, the resist comprises a fluorine-containing copolymer comprising a repeat unit derived from at least one ethylenically unsaturated compound characterized in that at least one ethylenically unsaturated compound is cyclic or polycyclic, at least one ethylenically unsaturated compound contains at least one fluorine atom covalently attached to an ethylenically unsaturated carbon atom, and at least one ethylenically unsaturated compound is comprised of a fluoroalcohol functional group having the structure:

-C (R<sub>f</sub>) (R<sub>f</sub>')OH

wherein  $R_f$  and  $R_f$ ' are the same or different fluoroalkyl groups of from 1 to about 10 carbon atoms or taken together are  $(CF_2)_n$  wherein n is 2 to 10. The fluorine-containing copolymer may contains sufficient functionality to render the photoresist developable so as to produce a relief image upon imagewise exposure to ultraviolet radiation having wavelength of < 365 nm.

In another embodiment, the resist comprises a fluorine-containing copolymer comprising:

- (i) a repeat unit derived from at least one ethylenically unsaturated compound containing at least three fluorine atoms covalently attached to two ethylenically unsaturated carbon atoms; and
- (ii) a repeat unit derived from an ethylenically unsaturated compound comprised of a fluoroalcohol functional group having the structure:

-C (R<sub>f</sub>) (R<sub>f</sub>') OH

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wherein  $R_f$  and  $R_f$ ' are the same or different fluoroalkyl groups of from 1 to about 10 carbon atoms or taken together are  $(CF_2)_n$  wherein n is 2 to 10. The fluorine-containing copolymer contains sufficient functionality to render the photoresist developable so as to produce a relief image upon imagewise exposure to ultraviolet radiation having wavelength of <365 nm.

In yet another embodiment, a photoresist comprises:

(a) a fluorine-containing copolymer comprising a repeat unit derived from at least one ethylenically unsaturated compound containing a fluoroalcohol functional group having the structure:

-XCH<sub>2</sub>C (Rf) (Rf') OH

wherein  $R_f$  and  $R_f$  are the same or different fluoroalkyl groups of from 1 to about 10 carbon atoms or taken together are (CF<sub>2</sub>) n wherein n is 2 to 10; and X is selected from the group consisting of sulfur. oxygen, nitrogen, phosphorous, other Group Va element. and other Group Vla element.

(b) at least one photoactive component; wherein the fluorine-containing copolymer contains sufficient functionality to render the photoresist developable so as to produce a relief image upon imagewise exposure to ultraviolet radiation having wavelength of < 365 nm.

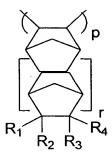
In certain embodiments, a photoresist comprises a fluorine-containing polymer comprising the structure:

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wherein each of  $R_1$ ,  $R_2$ ,  $R_3$ , and  $R_4$  independently is hydrogen, a halogen atom, a hydrocarbon group containing from 1 to 10 carbon atoms, a substituted hydrocarbon group, an alkoxy group, a carboxylic acid, a carboxylic acid ester, or a functional group containing the structure:  $-C(R_f)(R_f')$   $OR_b$  wherein  $R_f$  and  $R_f'$  are the same or different fluoroalkyl groups of from 1 to 10 carbon atoms or taken together are  $(CF_2)$  n wherein n is 2 to 10;  $R_b$  is hydrogen or an acid-or base-labile protecting group; p is the number of repeat units in the polymer; r is 0-4; at least one of the repeat units has a structure whereby at least one of  $R_1$ ,  $R_2$ ,  $R_3$ , and  $R_4$  contains the structure  $C(R_f)(R_f')$   $OR_b$ ; and wherein the fluorine-containing polymer has an absorption coefficient of less than 4.0 absorption unit per micron at a wavelength of 157 nm.

Fluorinated Alcohol (Co) polymers

A given fluorine-containing polymer or copolymer described above comprises a repeat unit (discussed infra) derived from at least one ethylenically unsaturated compound containing a fluoroalcohol functional group. These fluoroalkyl groups are designated as Rf and Rf', which can be partially fluorinated alkyl groups or fully fluorinated alkyl groups (i. e., perfluoroalkyl groups). Broadly, Rf and Rf' are the same or different fluoroalkyl groups of from 1 to about 10 carbon atoms or taken together are  $(CF_2)_n$  wherein n is 2 to 10. (In the last sentence, the terms "taken together" indicate that  $R_f$  and  $R_f$  are not separate, discrete fluorinated alkyl groups, but that

together they form a ring structure such as is illustrated below in case of a 5-membered ring:

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Rf and Rf can be partially fluorinated alkyl groups without limit according to the invention except that there should be a sufficient degree of fluorination present to impart acidity to the hydroxyl (-OH) of the fluoroalcohol functional group, such that the hydroxyl proton is substantially removed in basic media, such as in aqueous sodium hydroxide solution or tetraalkylammonium hydroxide solution.

In preferred cases according to the invention, there will be sufficient fluorine substitution present in the fluorinated alkyl groups of the fluoroalcohol functional group such that the hydroxyl group will have a pKa value as follows: 5 < pKa < 11. Preferably, Rf and Rf are independently perfluoroalkyl group of 1 to 5 carbon atoms, and, most preferably, R<sub>f</sub> and R<sub>f</sub> are both trifluoromethyl (CF<sub>3</sub>).

As is well known to those skilled in the polymer art, an ethylenically unsaturated compound undergoes free radical polymerization to afford a polymer having a repeat unit that is derived from the ethylenically unsaturated compound.

Specifically, an ethylenically unsaturated compound having structure:

$$P = C$$

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that undergoes free radical polymerization will afford a polymer having a repeat unit:

$$-\frac{P}{S}C-C\frac{Q}{T}$$

where P, Q, S, and T independently can be the same or different and illustratively could be fluorine, hydrogen, chlorine, and trifluoromethyl.

If only one ethylenically unsaturated compound undergoes polymerization, the resulting polymer is a homopolymer. If two or more distinct ethylenically unsaturated compounds undergo polymerization, the resulting polymer is a copolymer.

Some representative examples of ethylenically unsaturated compounds and their corresponding repeat units are given below:

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$$CF_3$$
 $C-CH_2$ 
 $C-CH_2$ 

wherein R is a hydrocarbon-containing group, preferably C<sub>1</sub> to C<sub>8</sub> alkyl.

Each fluorine-containing copolymer according to this invention has an absorption coefficient of less than 4.0 absorption units per micron at a wavelength of 157 nm, preferably of less than 3.5 absorption units per micron at this wavelength, more preferably, of less than 3.0 absorption units per micron at this wavelength, and, still more preferably, of less than 2.5 absorption units per micron at this wavelength.

The fluorinated polymers, photoresists, and processes that involve a fluoroalcohol functional group may have the structure:

# -XCH<sub>2</sub>C(R<sub>f</sub>)(R<sub>f</sub>') OH

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wherein  $R_f$  and  $R_f$  are the same or different fluoroalkyl groups of from 1 to about 10 carbon atoms or taken together are  $(CF_2)_n$  wherein n is 2 to 10; X is selected from the group consisting of oxygen, sulfur, nitrogen, phosphorous, other Group Va element, and other Group Vla element. By the terms "other Group Va element" and "other Group Vla element", these terms are understood to mean herein any other element in one of these groups of the periodic table that is other than the recited elements (i. e., oxygen, sulfur, nitrogen, phosphorous) in these groups. Oxygen is the preferred X group.

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Some illustrative, but non limiting, examples of representative comonomers containing a fluoroalcohol functional group and within the scope of the invention are presented below:

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$$CH_2C(CF_3)_2OH$$
  $CH_2OCH_2C(CF_3)_2OH$   $CH_2C(CF_3)_2OH$   $CH_2C(CF_3)_2OH$ 

CH<sub>2</sub>=CHOCH<sub>2</sub>CH<sub>2</sub>OCH<sub>2</sub>C(CF<sub>3</sub>)<sub>2</sub>OH

CH<sub>2</sub>=CHO(CH<sub>2</sub>)<sub>4</sub>OCH<sub>2</sub>C(CF<sub>3</sub>)<sub>2</sub>OH

$$OCH_2C(CF_3)_2OH$$
  $OCH_2C(CF_3)_2OH$ 

Various bifunctional compounds which can initially afford crosslinking and subsequently be cleaved (e. g., upon exposure to strong acid) are also useful as comonomers in the copolymers of this invention. As an illustrative, but nonlimiting example, the bifunctional comonomer NB-F-OMOMO-F-NB is desirable as a comonomer in the copolymers of this invention. This and similar bifunctional comonomers, when present in the copolymer component(s) of photoresist compositions of this invention, can afford copolymers that are of higher molecular weight and are lightly crosslinked materials. Photoresist compositions, incorporating these copolymers comprised of bifunctional monomers, can have improved development and imaging characteristics, since, upon exposure (which photochemically generates strong acid as explained infra), there results cleavage of the bifunctional group and consequently a very significant drop in molecular weight, which factors can afford greatly improved development and imaging characteristics (e. g., improve contrast).

Exposure at I-line (365 nm), DUV (248 – 254 nm, 193 nm), VUV (157 nm), EUV (aka x-ray 13.4 nm).

The invention encompasses negative-tone and positive-tone resists including, without limitation, those described herein.

Negative-tone resists are ones going from less polar to more polar by a wide range of chemistries known to those familiar in the art. The polarity change is typically brought about by an acid catalyzed reaction that causes a change in functional group. Negative-tone resists from a change in molecular weight, including crosslinking etc.

Positive-tone resists are ones going from a more polar group going to a less-polar group. Examples include, but not limited to, those set forth above and the pinacol rearrangement:

$$\begin{array}{c} \begin{array}{c} \\ \\ \\ \\ \\ \end{array} \end{array}$$

Synthesis of pinacol monomer:

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A number of photoacid generators can be used including, without limitation, those known to one skilled in the art. Examples of PAGs include triaryl sulfonium salts and diaryliodonium salts with perfluoroalkyl sulfonates

as the anion. Other anions included are those found in U.S. Patent No. 5,554,664, the disclosure of which is incorporated herein by reference in its entirety such as, without limitation, perfluoroalkyl sulfonyl imides and perfluoroalkyl sulfonyl methides. These PAGs are believed to have enhanced solubility characteristics and acidity.

Another example of a PAG which may be employed:

10 Preferably, R<sub>f</sub> is a fluoroalkyl substituent. It is preferred that the number of carbon atoms in the fluoroalkyl group range from 4 to 12, although other values are encompassed by the invention. Although not intending to be bound by any theory, it is believed that this range is optimal for controlled solubility and glass transition temperature (T<sub>g</sub>). Most preferably, the number of carbons is 6.

Preferably, R is either a fluoroalkyl substituent or an aryl-containing group (e.g., benzene or p-toluene). It is believed that the more acidic the sulfonic acid, the less stable the compound.

Examples of monomers and polymers that may be used include:

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$$R_1$$
 $R_2$ 
 $R_3$ 
 $R_4$ 
 $R_4$ 
 $R_4$ 
 $R_4$ 
 $R_4$ 
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 $R_4$ 
 $R_4$ 
 $R_4$ 
 $R_4$ 

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$R_1$	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>
F	F	F	F
F	F	F	$R_{\mathbf{f}}$
H	Н	Н	R <sub>f</sub>
$R_{f}$	Н	Н	$R_{\mathbf{f}}$
Н	Н	CH <sub>3</sub>	$COOR_f$
Н	Н	H	$COOR_f$
			$\begin{array}{c} \text{COOR}_{\text{f}} \\ \text{COOR}_{\text{f}} \\ \text{(CH}_2)_{\text{n}} \text{R}_{\text{f}} \text{ n=1,2} \end{array}$
R <sub>f</sub>	F	F	$R_{\mathbf{f}}$

wherein  $R_1$ ,  $R_2$ ,  $R_3$ , and  $R_4$  are defined in the table above. The above monomers may be used for copolymerization with other monomers disclosed herein. Preferred monomers for use in copolymerization include, without limitation, tetrafluoroethylene and hexafluoropropylene.

This invention addresses problems associated with image collapse, resist transparency at 157 nm, potential solubility problems of new resist materials in a readily available solvent, elimination of organic solvents and aqueous developers (i.e., turning a wet process into a dry process). The invention uses differences in polarity to create solubility differences in supercritical and liquid CO<sub>2</sub>.

The present invention may be used for the manufacture of semiconductor devices. In addition, the invention may be used for other non-lithographic processes in which a polymer must be deposited and removed from a surface in a controlled fashion (e.g., coating of low-k dielectrics, coating on solid substrates, fiber optics, optical components, glass, ceramics, metal, and plastics).

In the event that photoresist etch resistance difficulties may exist, such problems can be addressed by structure modification to make the polymers more resistant to etch. In the event that the polymers display adhesion problems, the CO<sub>2</sub> deposition process can help to potentially mitigate the difficulty in wetting the surface and reduce the chance of potentially catastrophic defects.

The invention has been described with respect to the preferred embodiments set forth above. It should be appreciated that these

embodiments are provided for illustrative purposes only, and do not limit the scope of the invention.

The invention has been described above in reference to its preferred embodiments. These embodiments do not serve to limit the scope of the invention as defined by the claims.